

Chromate (hexavalent) - Diphenylcarbazide Method

Version 7 | Mar 2018

Applications and Industries

Drinking water, surface and groundwater, domestic and industrial wastewater effluents

APHA Standard Methods, 22nd ed., Method 3500-Cr B -

ASTM D 1687- 02, Chromium in Water, Test Method A

Chemistry

Hexavalent chromium reacts with diphenylcarbazide under acidic conditions to form a red-violet colored complex in direct proportion to the hexavalent chromium concentration. Results are expressed as ppm (mg/L) CrO₄.

Available Analysis Systems

Visual colorimetric: CHEMets®

Instrumental colorimetric: Vacu-Vials®

Safety Information

Safety Data Sheets (SDS) are available upon request and at www.chemetrics.com. Read SDS before using these products. Breaking the tip of an ampoule in air rather than water may cause the glass ampoule to shatter. Wear safety glasses and protective gloves.

Storage Requirements

Products should be stored in the dark and at room temperature.

Shelf Life

When stored in the dark and at room temperature: Instrumental colorimetric:

Vacu-vials kit: at least 1 year

Visual colorimetric:

CHEMets refill, color comparators, Acidifier Solution: at least 1 year

Interference Information

The reaction with diphenylcarbazide is nearly specific for hexavalent chromium (chromate).

This chemistry does not measure trivalent chromium.

Permanganate and other oxidizing agents may oxidize trivalent chromium in the sample to hexavalent, causing a false positive result.

Sulfide, sulfite, and other reducing agents may reduce hexavalent chromium in the sample to trivalent, causing false low results.

Samples should be analyzed immediately upon collection to minimize the reactions of oxidizing and reducing

Samples with extreme pH or that are highly buffered (including samples preserved to pH 9.3 – 9.7 according to APHA Standard Methods Method 3500-Cr B-2009) may require pH adjustment. After addition of A-2800 Acidifier Solution, the pH of the sample should be 1.8 - 2.0. Up to 8 drops of A-2800 Acidifier Solution may be added to 20 mL of sample to adjust the pH to fall within this range.

Hexavalent molybdenum and mercury salts may react to form color with the reagent, but with much less sensitivity than does chromate.

Nitrite, as well as vanadium and titanium, may produce low test results.

Iron at greater than 1 ppm may produce a yellow color.

Accuracy Statement

CHEMets kit: + 1 color standard increment

Vacu-vials kit.

≤0.05 ppm at 0 ppm ±0.06 ppm at 0.20 ppm ±0.18 ppm at 0.90 ppm ±0.26 ppm at 2.60 ppm